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WO 02/05921 A1

(54) Title: CRYSTALLISATION PROCESS USING ULTRASOUND

(57) Abstract: The present invention relates to a process for the crystallisation of a solid phase from a liquid, characterised in that the liquid during crystallisation is subjected to ultrasound in the absence of transient cavitation. In particular the liquid is sonicated under such conditions of time and frequency that nucleation of stable crystals in the liquid is induced without exceeding the cavitation threshold and the occurrence of transient cavitation and the accompanying flavour deterioration is avoided. The liquid preferably is a triglyceride oil such as a vegetable oil or animal fat, e.g. liquefied milk fat.

5

CRYSTALLISATION PROCESS USING ULTRASOUND

The present invention relates to a process where a
liquefied or dissolved substance is crystallized from a
10 melt or a solution while exposing it to ultrasound. A
triglyceride fat (three fatty acid residues connected to a
glycerol backbone) in particular is the subject of the
present crystallisation process.

15 The triglyceride fats used for the manufacture of food
compositions often are desired to show a specific melting
behaviour. Fats as obtained from natural sources usually do
not have suitable melting properties. Therefore they have
to be subjected to a modification treatment. Fat
20 fractionation is such a modification treatment. Fat
fractionation consists of the physical separation of a
triglyceride mixture into two or more fractions with
different melting or solubility ranges. "Wet" fractionation
comprises dissolving the triglyceride mixture in a hot
25 organic solvent (e.g. hexane) and then cooling it slowly
until a part (fraction) of the fat crystallizes from the
solution.

Alternatively, "dry" fractionation does not make use of a
solvent and comprises cooling a liquid fat slowly.

30 Optionally a triglyceride mixture is first fully liquefied
if it is solid. The fat fraction with the highest melting
range will crystallize first during cooling.

The final stage of both wet and dry fractionation is
separation of the crystallized ("stearin") fraction and the
35 still liquid ("olein") fraction by filtration.

Dry fractionation is the preferred option when a "non-
chemical" modification treatment is desired. For dairy fats

5 it is the only acceptable option in terms of retaining flavour quality. However, dry fractionation is a less efficient and controllable method than wet fractionation (Ref.1).

The filter cake resulting from wet fractionation may
10 contain as little as 2 wt.% entrapped liquid fraction (also denoted as 98% SE (separation efficiency)). The good result is due to a more favourable crystal morphology and to washing the crystallized fraction with clean solvent. By contrast, the solids content in the cake resulting from a
15 standard dry fractionation process typically is at most about 60% (60% SE), the remaining 40% being entrapped olein.

Crystal habit modifiers (CHM's) when added to the melt modify the crystal morphology such that more compact
20 crystals may be produced which can be better separated from the liquid olein phase. The use of CHM's may increase the SE to about 80%, but at the expense of a much increased process time. CHM's slow down both nucleation and crystal growth. Moreover, for the removal of the CHM's from the
25 desired fat fractions additional post-processing is necessary.

Sonocrystallisation is the use of ultrasound for influencing the crystallisation of liquids, either melts or
30 solutions. Ultrasound in common language is sound characterized by a frequency of about 20 kHz and more, extending even into the MHz range. Most applications use ultrasound in the range 20 kHz - 5 MHz.

The >20 kHz frequency for defining ultrasound is rather
35 arbitrary and is historically related to the average perception limit of the human ear. Within the context of the present specification such perception limit is irrelevant from a technical point of view. The benefits of

5 the present invention become manifest as well with frequencies well below 20 kHz. In the context of the present specification ultrasound is defined as sound with a frequency of 10 kHz up to 10 MHz.

10 Since 1927 it is known that by exposing supercooled melts or supersaturated solutions of various substances to ultrasound the nucleation and/or the growth of crystals is remarkably influenced. The effect, sonocrystallisation, was first observed when crystallizing a supersaturated
15 thiosulfate solution. Since then sonocrystallisation has been studied in many other systems. A particular aspect of sonocrystallisation is sononucleation. It deals with the initiation of crystal formation, has been studied extensively with sugar and is applied since the late 50-
20 ties. Sonocrystallisation of supercooled water, supercooled metal melts and supersaturated solutions of various inorganic materials have received a lot of attention in the 50-ties and 60-ties, particularly in Russia.

25 The crystallisation process can be divided into two stages: crystal nucleation and crystal growth. In the nucleation stage submicroscopic crystal nuclei are formed which develop into larger crystals during the subsequent growth stage. With homogeneous nucleation the crystals are formed
30 directly from the liquid. Heterogeneous nucleation is nucleation mediated by foreign particles already present in the liquid. Secondary nucleation is nucleation mediated by pre-existing crystals. It is believed that the process of the present invention predominantly affects homogeneous
35 nucleation.

Benefits of sonocrystallisation reported in literature include:

WO 02/05921

- 5 • Faster nucleation which is fairly uniform throughout the sonicated volume,
- Relatively easy nucleation of materials for which nucleation is difficult otherwise,
- Generation of smaller, purer and more uniform crystals.
- 10 For literature dealing with sonocrystallisation see the reviews e.g. of Kapustin (Ref.2) and Hem (Ref.3).

When a liquid is exposed to ultrasound, microscopic gas/vapour bubbles are formed which show a dynamic pulsating behaviour. One activity of such ultrasound-

- 15 induced bubble behaviour is denoted as cavitation. Already at relatively low sound intensities the bubbles do not perish but exhibit stable volume and/or shape oscillations. This type of cavitation is denoted as "stable" or "non-inertial" cavitation. When the ultrasound intensity is increased and exceeds a certain limit, the cavitation threshold, the nature of cavitation changes dramatically which results in the bubbles becoming
- 20 unstable. Within a fraction of a sound cycle they show rapid growth followed by a violent collapse. The collapsing gas bubbles produce very high pressures and temperatures locally in the bubble as well as a high pressure in the liquid layer surrounding the bubble (see also Hem, 1967, supra).

- 25 Cavitation which shows this violent bubble behaviour is denoted as "transient" or "inertial" cavitation (ref.5). By many ultrasound users the terms "cavitation" and "transient cavitation" are used without discrimination.

- 30 According to general scientific consensus - which has persisted until now (see e.g. ref. 4 and 8) - the physical mechanism underlying sonocrystallisation and the benefits resulting from it are ascribed to the occurrence of

5 transient cavitation. The prejudice tells that in the
absence of transient cavitation the benefits of
sonocrystallisation even will not be manifested.

After the 60-ties the scientific attention for
10 sonocrystallisation seems to have decreased. No
fundamentally new insights in the believed underlying
cavitation mechanism have been reported. However, the
technological development and application of ultrasound for
the crystallisation of different materials continued.

15 A few patent applications relate to sonocrystallisation of
edible fats. WO 92/20420 describes a method and a device
for the control of solidification in liquids. The liquid to
be solidified is subjected to inter alia ultrasonic
20 cavitation in order to control the steps of nucleation
and/or crystal growth of the solidification process. In
conformity with prevailing views the ultrasonic conditions
desired for nucleation induction are chosen such that
transient cavitation results which implies high intensity
25 ultrasound.

EP 765605 deals with the effect of ultrasonic treatment on
fat nucleation. It describes a method for accelerating the
polymorphic transformation of edible fat compositions. Such
30 compositions when undercooled by at least 4°C are exposed
to ultrasonic energy for a time and at a frequency
sufficient to induce nucleation of stable polymorph
crystals without exceeding the melting point of those
crystals. Typical fats to be treated by this method are
35 butter fat and the fats used in ice cream, chocolate,
margarine and yogurt.

5 EP 765606 describes a method for retarding fat blooming on
chocolate and on other confectionery fat compositions
comprising cocoa butter. The method comprises undercooling
the molten fat by at least 3°C below the melting point of
the β -polymorph crystal. By exposing it to an effective
10 amount of ultrasonic energy stable crystals are generated.

In those patents cavitation is presented as the evident
cause of the enhanced nucleation and the changed crystal
morphology.

15

Traditional sonocrystallisation, however, has shown also
serious drawbacks. Sonocrystallisation may trigger
sonochemical reactions some of which are believed to cause
production of free radicals. Triglyceride fats, especially
20 unsaturated oils, are very susceptible to oxidation damage
caused by decomposition of lipo(hydro)peroxides formed by
free readicals. The resulting off-flavour and off-taste has
become a decisive factor preventing the wide use of
sonocrystallisation for edible unsaturated fats. A small
25 flavour defect in the predominantly saturated chocolate
fats as exemplified in the patents above is hardly noticed
and even less when incorporated in chocolate products.
Skilled fat chemists have persisted in believing that
sonocrystallisation of an unsaturated edible fat is
30 impossible without adversely affecting its taste and smell.

SUMMARY OF THE INVENTION

35 We have found that the beneficial effects of fat
sonocrystallisation being necessarily related to transient
cavitation are based on a prejudice.

5 The present inventors have found that sonocrystallisation
can considerably enhance the nucleation rate of fat
crystallisation also when applied in the absence of
transient cavitation. Adverse sonochemistry with its
flavour spoiling effects does not occur. A major
10 accomplishment was the significant improvement of the
separation efficiency of a dry fractionated oxidation
sensitive fat without the expected oxidation damage and
without adversely affecting the taste and smell of the
obtained fat fractions (see example 4).

15
Generally, the present invention provides a process for the
crystallisation of a solid phase from a liquid which liquid
is subjected to ultrasound, where the exposure to
ultrasound is at such conditions that transient cavitation
20 is absent and for a time and at a frequency sufficient to
induce nucleation of stable crystals in the liquid.

DESCRIPTION OF THE FIGURES

25
Fig.1. Is a diagram showing various applications of high
power ultrasound, ranging (along the Y-axis) from low to
high sound intensity and (along the X-axis) from
relatively low frequencies to high frequencies.

30
Fig. 2. Shows an experimental ultrasonic vessel component
assembly where various conditions which determine
cavitation can be varied.

35 Fig.3. Is a common mass spectograph showing characteristic
peaks of sonicated and non-sonicated sunflower seed oil
samples.

5 Fig.4. Depicts the time/temperature profiles of two fat blend samples during cooling.

Fig.5. Shows for a sonicated oil sample the single hydrophone signal at 1.5 MHz frequency and at 1.5 W/cm² intensity where besides the peak of the fundamental frequency no peaks of harmonics are visible. This hydrophone view is characteristic for the absence of transient cavitation.

15 Fig.6. Shows for a sonicated oil sample, in contrast to fig.5, the onset of subharmonics at 1.5 MHz where the sound energies have increased to such extent that the cavitation threshold has been exceeded.

20

DETAILS OF THE INVENTION

Generally, transient cavitation does not occur at low
25 ultrasound intensities. When the sound intensity is increased, eventually the transient cavitation threshold will be exceeded. As is discussed in several sources (see e.g. refs. 7 and 9), the occurrence of transient cavitation depends primarily on the intensity of the sound energy but
30 also on several other factors. The frequency of the ultrasound, the temperature and viscosity of the liquid, the amount of dissolved gas, and the presence of surface-active substances affecting the surface tension of the bubbles are the most important secondary factors. Fig. 1
35 illustrates the zones where for the various applications of ultrasound transient cavitation is likely to occur. The X-axis shows the sound frequency and the Y-axis the sound intensity. For applications situated in the top right

5 corner transient cavitation is always present, for applications shown in the bottom left corner cavitation is always absent. A generally applicable and sharply defined borderline for distinguishing the intensity threshold can not be given. However, in an operational situation with a
10 chosen frequency sound intensities where transient cavitation will not occur can be easily found with some trials. As will be discussed below for each operational situation indicators are available with which it is possible to distinguish whether sonication of a liquid finds
15 place in the presence or in the absence of cavitation. With the colloquial expression "subcavitation conditions" when used for sonication, the substantial absence of transient cavitation throughout the whole volume of crystallizing liquid is meant.

20

A practical indicator for the absence of transient cavitation is the value of the mechanical index (MI) of the actual ultrasound generating system. The MI is defined as

$$MI = (p_{NEG}[MPa]) / \sqrt{f[MHz]}$$

25 where $p_{NEG}[MPa]$ is the amplitude of the acoustic pressure of the ultrasound field (the pressure amplitude) and $f[MHz]$ is the ultrasound frequency. The MI is used as a risk indicator for indicating the worst-case likelihood of occurring inertial cavitation. It has been adopted by the
30 American Institute of Ultrasound in Medicine as a real-time output to estimate the potential risk of cavitation so that it can be avoided during diagnostic *in vivo* ultrasound scanning (ref. 5). According to Apfel and Holland (ref.7) transient cavitation does not occur when the MI of the
35 applied system does not exceed the threshold value 0.7. Hence, frequency and pressure amplitude of the ultrasound preferably is chosen such that said threshold value is not

- 5 exceeded. Since the sound intensity (I) is related to the pressure amplitude p_{NEG} according to the function

$$I = p_{\text{NEG}}^2 / 2\rho c$$

- the ultrasound intensity should not exceed the
10 corresponding intensity threshold value, where ρ (rho) is the liquid density and c the velocity of sound, which values in fat are about 920 kg/m³ and 1400 m/s respectively (and in water are hardly different).

- 15 The MI based threshold indicator is meant to distinguish riskless, medically safe sonication conditions from conditions where dangerous transient cavitation might, but not necessarily will occur. It precisely indicates the absence of transient cavitation, but less precisely
20 indicates the presence of transient cavitation.

- An alternative common and practical way for detecting the presence of transient cavitation is the observation of "sonoluminescence", which is the emission of very short
25 light flashes caused by collapsing cavitation bubbles in the presence of certain chemicals (ref.6). The method is not preferred, however, for clearly establishing the absence of transient cavitation.

- 30 Most suitably, however, the occurrence of transient cavitation can be detected by monitoring with a hydrophone the sound radiated by an ultrasonication cell. The hydrophone is a device which transforms sound energy emitted from a sonication cell into oscilloscope views. The
35 man skilled in the art of reading such views, will easily recognize the onset of transient cavitation by the appearance of peaks of characteristic harmonics and subharmonics and eventually the appearance of "noise" which

5 belongs to full cavitation. The harmonics and sub-harmonics
result from the non-linear volume oscillations of strongly
driven cavitation bubbles. The shock waves produced by
imploding bubbles become visible because they create broad-
band pulses in the frequency spectrum. The superposition of
10 many such signals from all bubble implosions generated by a
cavitating sound field gives rise to a broad-band "noise"
signals pattern. Hence, such noise pattern points to the
many violent bubble collapses which are characteristic for
transient cavitation. By contrast, bubble oscillations
15 during stable, non-transient cavitation do not show a noise
pattern in the hydrophone view (ref. 9).

The sonocrystallisation process of the present invention
employs such low intensity ultrasound that a hydrophone,
20 when detecting sound radiated from the ultrasound exposed
liquid, shows a signals pattern which is free from broad-
band cavitation noise.

A preferred embodiment of the present invention is
25 characterized by the ultrasound intensity being at such low
level that a hydrophone when detecting sound radiated from
the ultrasound exposed liquid shows a view with a main
signal corresponding with the main radiation frequency and
a further signal corresponding with the first subharmonic
30 frequency where the intensity peaks ratio of the further
signal and the main signal, the peaks ratio A_s/A_F , is < 0.5 .

Most preferably the ultrasound intensity is at such low
level that a hydrophone when detecting sound radiated from
35 the ultrasound exposed liquid shows a view with a single
signal corresponding with the main radiation frequency
without substantially showing additional signals

5 corresponding with subharmonics frequencies.

It should be noted that the claimed condition "In the absence of transient cavitation" includes conditions with the occasional occurrence of transient cavitation. Such
10 occasional cavitation does not give rise to the noise pattern as detectable by a hydrophone and equally will not have an adverse effect on the sensoric properties of the treated fat.

15 It should be further noted that the intensity of the energy radiating from an ultrasound probe is fading away with an increasing distance from the energy source. At a relatively large distance from the probe cavitation is always absent. In a large volume of liquid cavitation may occur near the
20 ultrasonic probe while at the same time cavitation is absent at remote places of the same liquid. Therefore the criterion of the present invention is that transient cavitation is absent throughout the whole volume of the sonicated liquid.

25

Processing conditions other than the ultrasound intensity such as time and temperature and frequency as mentioned before can easily be optimized by the skilled person by some trials. It has been found, e.g., that for ultrasound
30 crystallisation of anhydrous milk fat the intensity optimum is just below the cavitation threshold (example 4). Generally, a too long exposure of the crystallized fat to ultrasound may cause a collapse of the crystal structure. Sonocrystallisation is particularly effective when cooling
35 has proceeded so far that the system has become supersaturated.

5 In principle, the present invention is suitable for the
sonocrystallisation of all kinds of liquids. It has been
found to be particularly useful for sonocrystallisation of
triglyceride oils either being of vegetable or of animal
origin or being a mixture of both. Preferably, the
10 triglyceride oil is of vegetable origin and is selected
from the group consisting of rapeseed oil, palmkernel oil,
sunflower oil, groundnut oil, mustard oil, safflower oil,
sesame oil, corn oil, soybean oil, cottonseed oil, linseed
oil and olive oil. Oils having an animal origin include
15 marine oils and milk fat. All those fats are more or less
unsaturated and are susceptible for adverse sonochemistry
and flavour deterioration when treated by traditional
sonocrystallisation.

Some fats are solid at ambient temperature and have to be
20 liquefied by heating before a dry fractionation process can
be carried out. Most of the mentioned vegetable fats are
liquid and do not need an initial liquefying step.

Preferably the fats are unmodified, but also modified fats
25 such as hydrogenated fats or fats which have been subjected
to interesterification will benefit from the present
invention.

A preferred embodiment of the present invention is a
30 process for fractionating a triglyceride fat, which
comprises the steps of:

- a. when the fat is solid, heating the triglyceride fat
until no substantial amount of solid triglyceride fat is
present in the oil,
- 35 b. allowing the triglyceride oil to cool and to
crystallize resulting in a solid stearin fraction and a
liquid olein fraction,

- 5 c. recovering the stearin fraction by separating it from the olein fraction, characterised in that during step b. the oil is exposed to ultrasound in the absence of transient cavitation.
- 10 A typical vessel suited for batch fractionation is equipped with proper means for heat exchanging, for stirring the vessel content, for applying ultrasound energy and for monitoring the occurrence of cavitation. It goes without saying that alternative equipments can be arranged with
- 15 devices which equally will allow the invention to be carried out. The sonication vessel could be filled via a pre-cooling unit; the sonication being started either in that unit or in the tube conducting the liquid to be crystallized to the main crystallisation vessel.
- 20 Other embodiments of the invention relate to processes for the preparation of edible emulsion spreads which may be either water continuous or fat continuous. The most common spreads such as margarine have a continuous fat phase and a
- 25 dispersed aqueous phase. Such spreads are traditionally prepared by passing a mixture of the aqueous phase and the oil phase through a series of one or more scraped-surface heat exchangers and pin stirrers. The oil phase of those mixtures is eventually crystallized by cooling under such
- 30 shear that a plastic W/O-emulsion is obtained in which a lattice of fine fat crystals provides the desired consistency and stabilizes the dispersed aqueous phase.
- Alternatively the process of spread preparation may start
- 35 with a continuous aqueous phase emulsion and includes a phase inversion step in order to impart fat continuity to the emulsion spread.

5 The lattice of fat crystals in the spread necessarily consists of solid saturated fat. For reasons of healthy nutrition and economy of raw materials the content of such saturated fat preferably is restricted to the minimal functional amount. The present invention has shown to have
10 such a beneficial influence on nucleation and eventually on the strength of the crystal lattice that even at relatively low solid fat levels a spread product with a good consistency, texture and stability is obtained.

15 Consequently the present invention provides a process for the preparation of a fat continuous emulsion spread comprising the steps of

- a. mixing a liquefied fat phase comprising essentially no solid fat and an aqueous phase so that a water-in-oil
20 emulsion results,
- b. cooling and working the emulsion to cause partial crystallisation of the fat until a desired consistency and texture is obtained,
characterised in that in the step comprising fat
25 crystallisation the emulsion is exposed to ultrasound in the absence of transient cavitation.

Alternatively, the present invention further provides a process for preparing a W/O-emulsion spread comprising the
30 steps of

- a. preparing an O/W-emulsion having a continuous aqueous phase containing dispersed fully liquefied fat,
cooling the emulsion to cause partial crystallisation of the fat, so obtaining a dispersion of partially
35 crystallized fat in a continuous aqueous phase,
- b. inverting the O/W-emulsion into a fat continuous emulsion in the usual way,

- 5 c. working and cooling the fat continuous emulsion to
cause further partial crystallisation of the fat until
a desired consistency and texture is obtained,
characterized in that in the step comprising fat
crystallisation the emulsion is exposed to ultrasound in
10 the absence of transient cavitation.

For present spread manufacturing processes the invention is
most beneficial for the preparation of emulsion spreads
which are fat continuous. Proper fat crystallisation plays,
15 however, also a role in the preparation of spreads in which
fat is the dispersed phase and where sonocrystallisation
according to the present invention also is a most
beneficial tool.

- 20 A since long acknowledged benefit of sonocrystallisation is
its potential influence on the habitus of the crystallized
fat. The formation of one fat polymorph may be promoted
over another one. Since some polymorphs possess preferred
properties, sonocrystallisation provides a tool for
25 improving the properties of the resulting fat and
indirectly for improving the properties of food products
containing those triglyceride fats.

It should be noted that the invented sonication treatment
30 is a new tool for fat modification that creates the chance
but not the guarantee of improved nucleation or of the
formation of a SE enhancing crystal morphology.

Processes, ingredients and equipment for fat fractionation
35 and for the preparation of said emulsion spreads, the fat
continuous as well as the water continuous ones, are well
known by the person skilled in the art and can be found
with all details in various textbooks such as K.A.

- 5 Alexandersen, Margarine Processing Plants and Equipment
(Vol.4, Bailey's Industrial Oil and Fat Products, Wiley and
Sons Inc., New York 1996).

EXAMPLES

10

Besides a commercial ultrasound probe geared to generate transient cavitation sound, we used for the exposure of the following exemplified samples to ultrasound the experimental device as illustrated by Fig. 2.

- 15 It comprises a vessel 1 comprising an inner perspex jacket 2 and an outer perspex jacket 3. The vessel 1 is generally cylindrical and closed at both ends. A thermocouple arrangement 4 projects into the body of vessel 1 through one of the ends. The thermocouple is combined with a
20 hydrophone arrangement to monitor the emitted ultrasound.

At the other end of vessel 1 cooling/heating coils and also a blade stirrer project into the body of the vessel.

- For generating ultrasound two circular transducers 5 and 6
25 are located circumferentially around the periphery of the inner perspex jacket 2. These are held in place by alignment rings 7, 8, 9 and 10.

- The ultrasound is generated and controlled by readily available standard equipment. It adjusts the frequency and
30 intensity of the ultrasound as appropriate.

- The installed transducer is capable of operating both below and above the cavitation intensity threshold. The cell is further provided with means for controlling the temperature of the sample and for delivering the sound energy either
35 continuously or pulse-wise.

While monitoring the hydrophone the frequency of the ultrasound in the device of Fig.2 is adjusted such that a

- 5 suitable resonant ultrasound frequency is found and maintained. Particularly the 10-11 kHz region is suited.

Example 1

10

This example and the next one are meant to compare sonocrystallisation of triglyceride oil samples with and without transient cavitation and to show that cavitation induced sonochemistry is actually related to the occurrence of off-flavours.

15

The test uses ultrasound generated by a common commercial Branson™ probe. Like the majority of industrial ultrasonic probes it is meant to produce high intensity fields at relatively low frequencies so that the believed beneficial cavitation bubble clouds are generated in the exposed material. A high intensity sound energy burst is emitted at a frequency of 20 kHz.

20

Refined sunflower oil was exposed to ultrasound using the lowest power output of 30 W of this ultrasound device, the exposure time varying from 1 to 10 minutes. The sonication cell was maintained at 20°C and samples were stirred at a constant rate. For detecting the expected sonochemical changes mass spectroscopy was used as the instrumental method, supplemented with sensoric sniffing of the samples (see also example 2).

30

Fig 3 shows the mass spectrum for both the sunflower oil sample sonicated under cavitation conditions and a comparison non-sonicated sample. Several of the ultrasound-induced mass spectrum peaks of the sonicated sample were recognized as related to known

35

5 off-flavour compounds by the scientists skilled in testing oils on deterioration. The oil deterioration was further confirmed by a sensory panel test (see also example 2).

When investigating the deterioration effects of
10 sonocrystallisation on triglyceride oil as a function of the sound intensity, frequency, temperature, presence of oxygen, addition of water, metal-ion contamination and storage conditions, it has appeared that the major cause for off-flavour formation was the occurrence of cavitation
15 during sonication.

Example 2

The present example compares triglyceride
20 sonocrystallisation with and without transient cavitation and shows the findings of a sensory panel on the formation of off-flavour. A bland refined sunflower oil was divided in four samples A, B, C and D. Sample B was the only one not sonicated .

25 Each of the samples A, C and D was at 20°C subjected in a sonication cell to the sonication conditions A, C and D:

A. The oil was sonicated for only 3 minutes using the common Branson™ probe which is meant to generate transient
30 cavitation.

C. The oil was sonicated for one hour in the device of Fig.2 at a sound intensity near the cavitation threshold. Only occasionally transient cavitation occurred.
35

D. The oil was sonicated for one hour in the device of Fig.2 at a sound intensity for which not any transient

5 cavitation could be observed.

The occurrence of transient cavitation was monitored using a hydrophone.

10 Each of the three sonicated samples was submitted to a sensoric panel (n=22) for flavour assessment. Each panel member received successively each of the three samples accompnied by the untreated sunflower sample B without knowing which of both was the untreated sample. Each panel
15 member had to answer the question whether (s)he could perceive a flavour/taste difference between both samples. Table I summarizes the panel response.

TABLE I

20

Sonicated Oil Panel Assessment (n=22)			
Sunflower Oil (SF) Sample	No differ ence	Hesita tion or slight differ ence	Clear differ ence
Bland SF not sonicated (comparison)			
SF 3 min sonicated with transient cavitation (1)	0	0	22 (3)
SF 1 hour sonicated with occasionally occurring transient cavitation (2)	17	5	0
SF 1 hour sonicated without any transient cavitation (2)	19	3	0

(1) Branson™ sonication probe used

(2) Device of figure 2 used

(3) Flavour characterized as: metal, fishy, off

25

The experiment made clear that an unsaturated oil as sunflower seed oil can be subjected to an ultrasound treatment without substantial damage to flavour and taste.

5

Example 310 Sonocrystallisation in the absence of transient cavitation

This example demonstrates that in contrast to general belief also in the absence of transient cavitation sononucleation can be demonstrated.

15

This time the chosen sound frequencies are in the MHz area which are common for medical applications (see Fig.1).

A blend of 12% hydrogenated palm oil dissolved in 88%
20 sunflower oil of 60°C was divided in two samples (a) and (b). Both were poured into an ultrasound cell according to figure 2 and continuously cooled. From 45°C downwards sample (b) was exposed to continuous 1.5 MHz ultrasound at an intensity of 1.5 W/cm². Sample (a) was cooled in the same
25 way but without sonication.

Fig.4 shows the temperature graphs of both samples during the cooling period.

After 50 minutes of cooling a sudden temperature rise in sample (b) occurred which is ascribed to the release of
30 heat of crystallisation at the onset of fat crystallisation. Ten minutes after the occurrence of that peak the sample became turbid of fat crystals. At that time sample (a) did not yet show any fat crystallisation.

35 The sonication was monitored with a hydrophone which only had shown (Fig.5) the single peak 1.5 MHz peak of the ultrasound sound frequency which means that transient cavitation had been absent.

5 The MI value being 0.09, is far below the transient cavitation threshold of 0.7, which further confirms the absence of cavitation.

Fig.6 shows in contrast with Fig.5 a hydrophone view of high intensity ultrasound sonication where the cavitation
10 threshold had been exceeded. That transient cavitation prevails is apparent from the various of (sub)harmonics peaks.

15

Example 4

Fat fractionation with the use of ultrasound

Butterfat (AMF, anhydrous milk fat) was obtained from
20 Corman. The fat was melted and, while stirring (50 rpm), was kept at 65°C for at least 1 hour to ensure thorough melting and to avoid so-called "memory effect".

Subsequently it was cooled to 40°C in one hour and then to 33°C at 5°C/h. Only after the final temperature was reached
25 sonication without transient cavitation was applied on the supersaturated sample for 15 minutes (65 kHz, 30 dB). Then the sample was kept overnight at 33°C without stirring to let the crystallisation process proceed to completion.

The crystals were vacuum filtered (factor 3 ceramic filter)
30 for 30 minutes and then pressed. The pressure was gradually increased to 12 bar over a period of 60 minutes.

The anhydrous milk fat (AMF) commonly is dry fractionated with a separation efficiency of 60%. Table II shows that
35 use of ultrasound gave a SE of 80%, a spectacular improvement over the control. The hydrophone at no time showed the occurrence of transient cavitation. The flavour quality of the crystallised fat was not affected.

TABLE II

Sample	Dispersion Solids %	Filtered Solids %	Pressed Solids % (12 bar)
Control 1	8.7	18.7	60.7
Control 2	9.1	22.1	60.8
15 min at 33° C 1	7.1	18.7	78.8
15 min at 33° C 2	8.1	21.0	82.3

- 10 The experiment was repeated with other ultrasonic intensities, but all in the absence of transient cavitation. The intensity optimum for sononucleation appeared to be just below the cavitation threshold.
- 15 Ultrasound caused a dramatic effect on crystal size, shape and distribution. The textures of the final fat fractions appeared to be very different from each other as well as from the non-sonicated control sample. This example proves that kinetics and structure of fat crystals may be greatly
- 20 affected by exposure to ultrasound even in the absence of transient cavitation.

REFERENCES

1. Hamm, Trans. IChemE. 74C, 1996, 61.
- 10 2. Kapustin, The effects of ultrasound on the kinetics of crystallisation; Consultants Bureau, New York, 1963.
3. Hem, The effect of ultrasonic vibrations on crystallisation processes; Ultrasonics, October 1967, p.
15 202.
4. Kallies, Zur gezielten Suspensionserzeugung für die Konfektionierung von Schmelzen; PhD Thesis, Bremen 1995.
- 20 5. Leighton T.G., The principles of cavitation; chapter 9 in "Ultrasound in Food Processing", (Povey & Mason, eds.) Blackie Academic & Professional, London (1998).
6. Crum L.A., Sonoluminescence, sonochemistry and
25 sonophysics, J. Acoust. Soc. Am. **95**(1), 1994, 559.
7. Apfel R.E., Holland C.K., Gauging the likelihood of cavitation from short-pulse low-duty cycle diagnostic ultrasound, Ultrasound Med.Biol. **17**, 1991, 179.
- 30 8. Internetsite <http://www.aeat.com/sono/>, particularly "How does it work" under "Frequently Asked Questions". Printed on 21 March 2001.
- 35 9. Gélat P., Hodnett M, Zeqiri B., Establishing a reference ultrasonic cleaning vessel: part 1: Supporting infrastructure and early measurements, National Physical Laboratory Report CMAM 55, September 2000.

CLAIMS

1. Process for the crystallisation of a solid phase from a liquid which liquid is subjected to ultrasound, characterized in that the exposure to ultrasound is at such conditions that transient cavitation is absent and for a time and at a frequency sufficient to induce nucleation of stable crystals in the liquid.
2. Process according to claim 1, characterized in that ultrasound intensity is adjusted at such low level that a hydrophone when detecting sound radiated from the ultrasound exposed liquid shows a view which is free from broad-band cavitation noise signals pattern.
3. Process according to claims 1 or 2, characterized in that ultrasound intensity is at such low level that a hydrophone when detecting sound radiated from the ultrasound exposed liquid shows a view with a main signal corresponding with the main radiation frequency and a further signal corresponding with the first subharmonic frequency where the intensity peaks ratio of the further signal and the main signal is < 0.5 .
4. Process according to any one of claims 1 to 3, characterized in that ultrasound intensity is adjusted at such low level that a hydrophone when detecting sound radiated from the ultrasound exposed liquid shows a view with a single signal corresponding with the main radiation frequency without

substantially showing additional signals corresponding with subharmonics frequencies.

5. Process according to any one of claims 1 to 4 ,
characterized in that an ultrasound generating system is
used of which the mechanical index (MI) is < 0.7 , where

$$MI = (p_{NEG}[MPa]) / \sqrt{f[MHz]}$$

and where $p_{NEG}[MPa]$ is the amplitude of the acoustic pressure
of the ultrasound field (the pressure amplitude) and $f[MHz]$
is the ultrasound frequency.

6. Process according to anyone of the previous claims,
characterized in that the liquid is a triglyceride oil of
vegetable or animal origin or a mixture of both.
7. Process according to claim 6, characterized in that the
triglyceride oil of vegetable origin is selected from the
group consisting of rapeseed oil, palmkernel oil, sunflower
seed oil, groundnut oil, mustard oil, safflower oil, sesame
oil, corn oil, soybean oil, cottonseed oil, linseed oil and
olive oil.
8. Process according to claim 6, characterized in that the
triglyceride oil is a liquefied dairy fat.
9. Process for fractionating a triglyceride oil, which
comprises the steps of:
- a. when the fat is solid, heating the triglyceride oil until
no substantial amount of solid triglyceride is present in
the oil;
 - b. allowing the triglyceride oil to cool and to crystallize

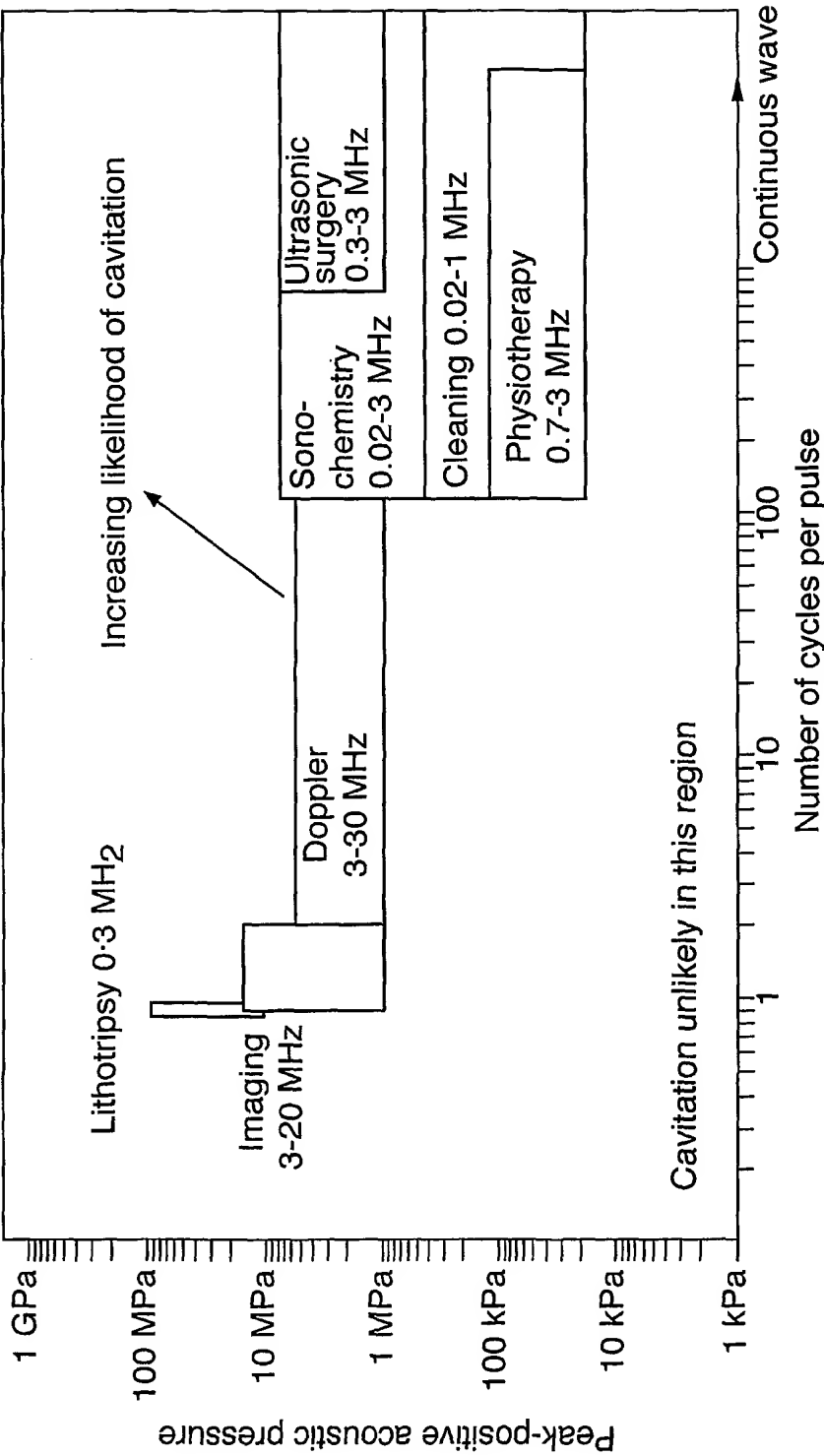
- resulting in a solid stearin fraction and a liquid olein fraction;
- c. recovering the stearin fraction by separating it from the olein fraction,
- characterised in that during step b. the oil is exposed to ultrasound in the absence of transient cavitation.
10. Process for the preparation of a fat continuous emulsion spread comprising the steps of
- a. mixing a liquefied fat phase comprising essentially no solid fat and an aqueous phase so that a water-in-oil emulsion results;
- b. cooling and working the emulsion to cause partial crystallisation of the fat until a desired consistency and texture is obtained,
- characterised in that in the step comprising fat crystallisation the emulsion is exposed to ultrasound in the absence of transient cavitation.
11. Process for preparing a W/O-emulsion spread comprising the steps:
- a. preparing a O/W-emulsion having a continuous aqueous phase containing dispersed fully liquefied fat cooling the emulsion to cause partial crystallisation of the fat, so obtaining a dispersion of partially crystallized fat in a continuous aqueous phase;
- b. inverting the O/W-emulsion into a fat continuous emulsion,
- c. working and cooling the fat continuous emulsion to cause partial crystallisation of the fat until a desired consistency and texture is obtained,
- characterized in that in the step comprising fat

crystallisation the emulsion is exposed to ultrasound in the absence of transient cavitation.

12. Process for preparing a O/W-emulsion spread comprising the steps:

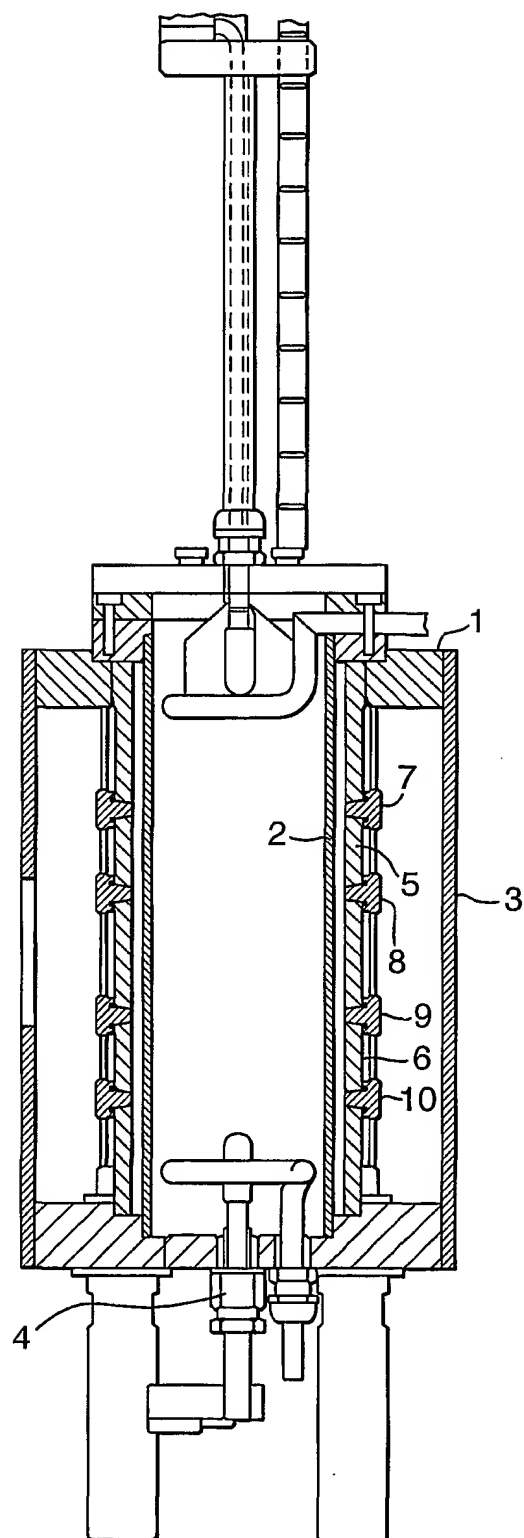
- a. preparing a O/W-emulsion having a continuous aqueous phase and a dispersed fully liquefied fat phase and cooling the emulsion to cause partial crystallisation of the fat, so obtaining a dispersion of partially crystallized fat in a continuous aqueous phase;
- b. working and cooling the fat continuous emulsion to cause partial crystallisation of the fat until a desired consistency and texture is obtained, characterized in that in the step comprising fat crystallisation the emulsion is exposed to ultrasound in the absence of transient cavitation.

Fig.1.



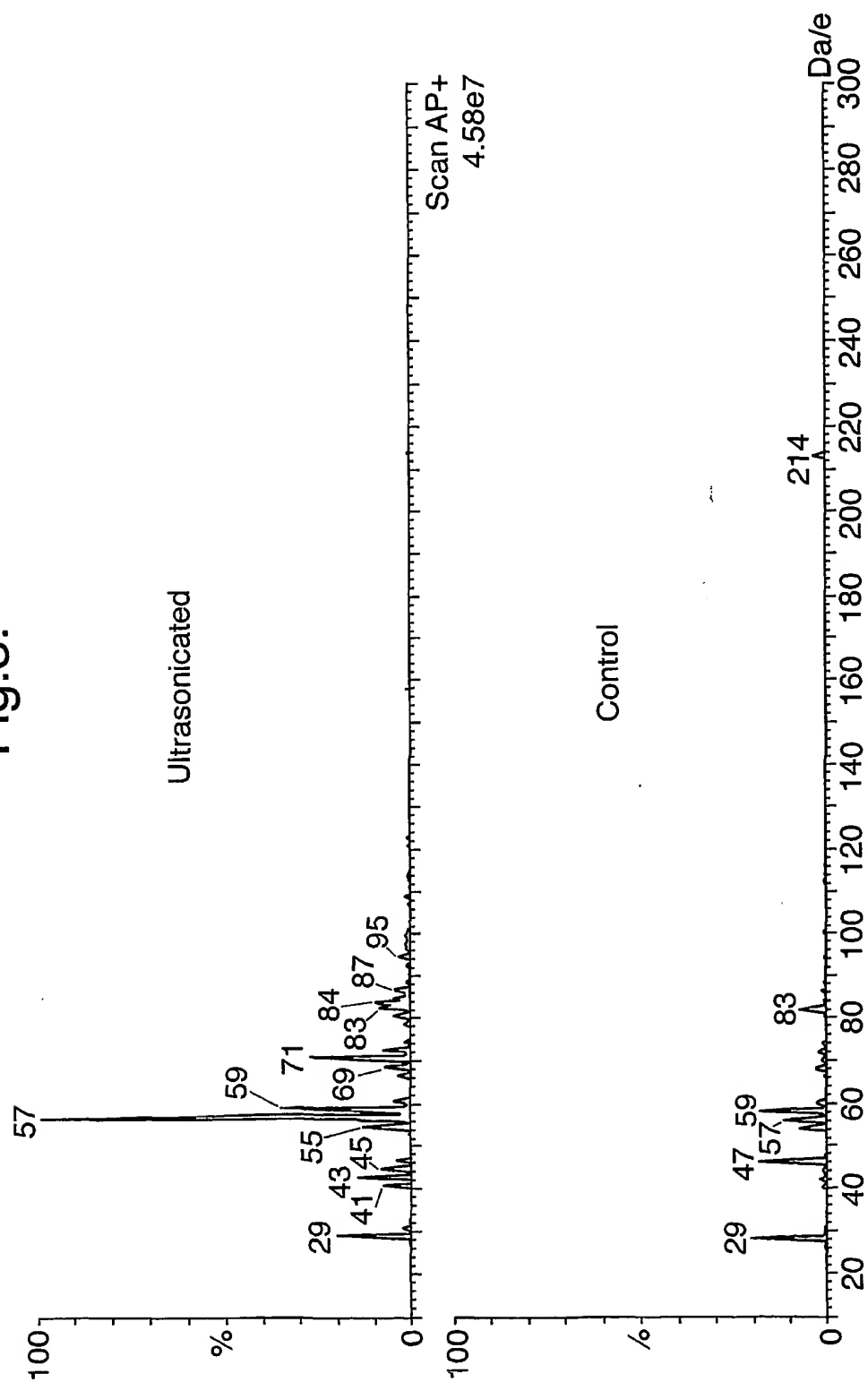
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Fig.2.



3/6

Fig.3.



4/6

Fig.4.

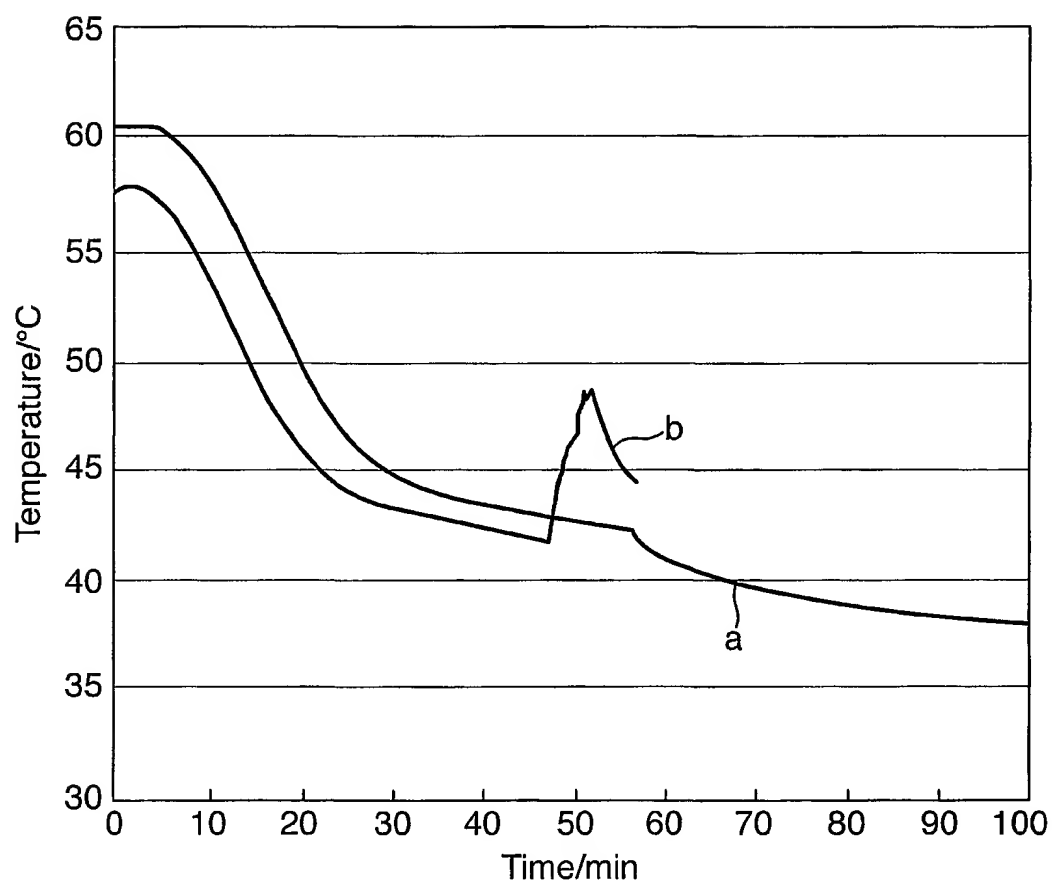


Fig.5.

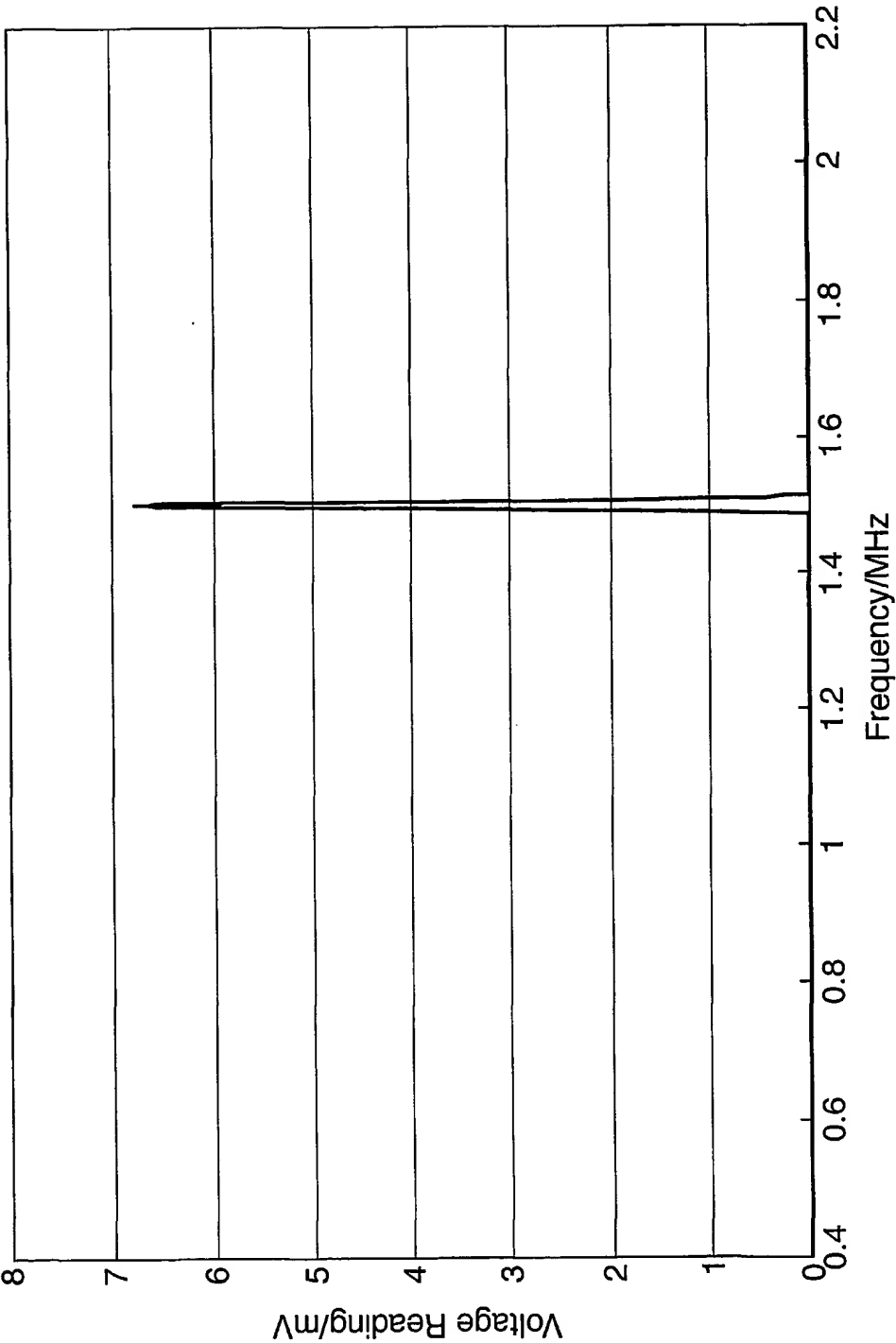
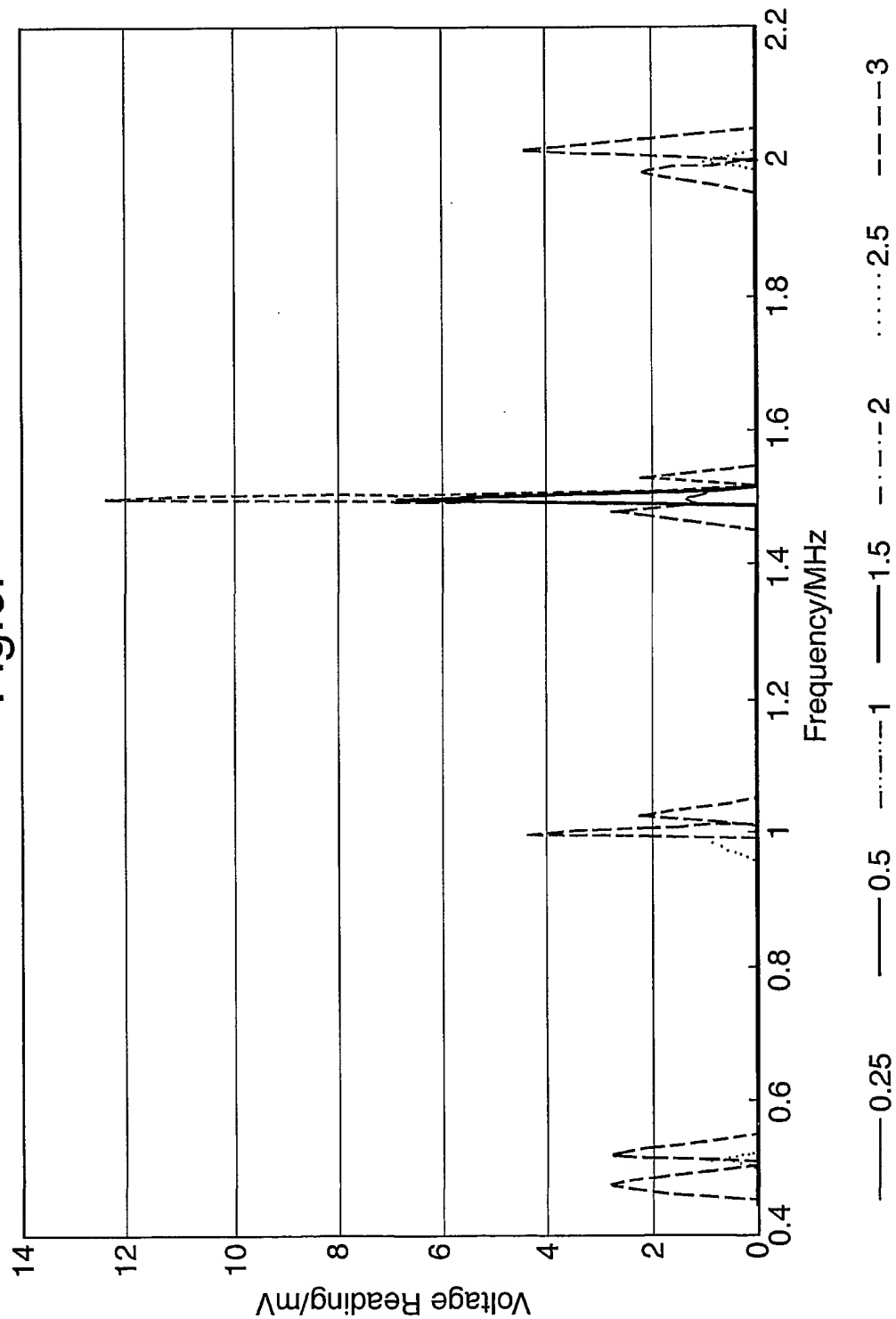


Fig.6.



INTERNATIONAL SEARCH REPORT

International Application No

PCT/EP 01/08022

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 B01D9/00 C11B7/00 C11B15/00 A23D7/05

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 B01D C11B A23D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ, INSPEC, COMPENDEX

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 765 605 A (KRAFT JACOBS SUCHARD R & D INC) 2 April 1997 (1997-04-02) the whole document ---	1-4
A	WO 92 20420 A (ACTON ELIZABETH ;MORRIS GEORGE JOHN (GB)) 26 November 1992 (1992-11-26) cited in the application the whole document ---	1-4
A	EP 0 619 139 A (ATOMIC ENERGY AUTHORITY UK) 12 October 1994 (1994-10-12) page 2 ---	1
A	US 5 209 879 A (REDDING JR BRUCE K) 11 May 1993 (1993-05-11) column 9, line 46 -column 10, line 44 claims --- -/--	1

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PC. 01/08022

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 4 594 194 A (DIEFFENBACHER ALBRECHT) 10 June 1986 (1986-06-10) the whole document	7
A	US 4 438 149 A (VERHAGEN LAURENTIUS A M ET AL) 20 March 1984 (1984-03-20) the whole document	8
A	EP 0 613 620 A (DAIRYGOLD TECH LTD) 7 September 1994 (1994-09-07) the whole document	9

INTERNATIONAL SEARCH REPORT

Int. Application No.

PCT/JP 01/08022

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
EP 0765605	A	02-04-1997	EP 0765605 A1	02-04-1997
WO 9220420	A	26-11-1992	AU 1673692 A	30-12-1992
			DE 69218194 D1	17-04-1997
			DE 69218194 T2	02-10-1997
			EP 0584127 A1	02-03-1994
			WO 9220420 A1	26-11-1992
			JP 6509498 T	27-10-1994
			KR 215017 B1	16-08-1999
EP 0619139	A	12-10-1994	DE 69410520 D1	02-07-1998
			DE 69410520 T2	24-09-1998
			EP 0619139 A1	12-10-1994
			GB 2276567 A ,B	05-10-1994
			JP 7000810 A	06-01-1995
			US 5395593 A	07-03-1995
US 5209879	A	11-05-1993	AU 7687291 A	30-10-1991
			CA 2079916 A1	07-10-1991
			CN 1055950 A	06-11-1991
			EP 0525039 A1	03-02-1993
			US 5460756 A	24-10-1995
			WO 9115307 A1	17-10-1991
US 4594194	A	10-06-1986	CH 658163 A5	31-10-1986
			CA 1214064 A1	18-11-1986
			DE 3471495 D1	30-06-1988
			EP 0139177 A1	02-05-1985
			GB 2147605 A ,B	15-05-1985
			IN 158233 A1	27-09-1986
			JP 60101197 A	05-06-1985
			PH 22071 A	20-05-1988
			SG 66287 G	04-03-1988
US 4438149	A	20-03-1984	NL 8101639 A	01-11-1982
			AT 7843 T	15-06-1984
			AU 546889 B2	26-09-1985
			AU 8216182 A	07-10-1982
			CA 1175284 A1	02-10-1984
			DE 3260240 D1	19-07-1984
			EP 0063389 A1	27-10-1982
			FI 821128 A ,B,	03-10-1982
			FR 2502903 A1	08-10-1982
			IE 53028 B1	11-05-1988
			JP 1467096 C	30-11-1988
			JP 57177650 A	01-11-1982
			JP 62018130 B	21-04-1987
			ZA 8202299 A	30-11-1983
EP 0613620	A	07-09-1994	EP 0613620 A2	07-09-1994
			IE 940189 A2	07-09-1994